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SECTION 3 continued 4. Table 6on page 485 of the text lists several amphoteric species, but only one other than water is neutral. NH 3 a. Identify that neutral compound. b. Write two equations that demonstrate this compound's amphoteric properties. (Answers will vary, but one equation should form NH

14 Acids and Bases
3(g) 28.0 g/mol N 2 a. Determine to one decimal place the molar mass of each substance and express each mass in grams per mole. 2.0 g/mol H 2 17.0 g/mol NH 3 b. There are six different mole ratios in this system. Write out each one. 3 mol H 2:1 mol N 2; 2 mol NH 3:1 mol N 2; 2 mol NH 3:3 mol H 2; or their reciprocals 7. Given the following ...

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SECTION 1 continued Date Class /a/3Qs— 20/3 5. Convert a pressure of 0.200 atm to the following units: 152 20.3 a. mm Hg b. kPa Of ROC) 6. When an explosive like TNT is detonated, a mixture of gases at high temperature is created. Suppose that gas X has a pressure of 50 atm, gas Y has a pressure of 20 atm, and gas Z has a pressure of 10 atm ...

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MODERN CHEMISTRY SECTION 2-3 REVIEW 13 HRW material copyrighted under notice appearing earlier in this work. SECTION 2-3 continued 4. Carry out the following calculations. Express each answer to the correct number of significant figures. a. 37.26 m 2.7 m 0.0015 m b. 256.3 mL 2 L 137 mL c. d. 5.