

Calorimetry Lab Gizmo Answers

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Calorimetry Lab Gizmo Explore Learning Answer Key - PDF ...

Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper.

Student Exploration- Calorimetry Lab (ANSWER KEY)

Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows

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temperatures of the individual substances after mixing.

Calorimetry Lab Gizmo : ExploreLearning

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This project was created with Explain Everything™ Interactive Whiteboard for iPad.

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Launch Gizmo Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water.

Calorimetry Lab Gizmo : Lesson Info : ExploreLearning

Calorimetry Lab In this Gizmo, you will investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. ****YOU WILL NEED YOUR OWN GIZMO ACCOUNT TO COMPLETE THIS ACTIVITY. INSTRUCTIONS ARE HERE** (be sure to record your login info on Pg G of your notebook)

Gizmo: Calorimetry - Instructure

Correct Answer: C. The final temperature of the lead-water

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system will be lower than the final temperature of the copper-water system. A blacksmith heats a 1,540 g iron horseshoe to a temperature of 1445°C before dropping it into 4,280 g of water at 23.1°C.

Calorimetry Lab Flashcards | Quizlet

Student Exploration: Calorimetry Lab Vocabulary: calorie, calorimeter, joule, specific heat capacity Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. The Latin word calor means “heat,” and meter comes from the Greek word meaning “to measure.” What do you think a calorimeter does? Measures heat 2. Where have you heard the word calorie before? ? What do you think a calorie

Calorimetry Lab SE - Student Exploration Calorimetry Lab

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance’s specific heat capacity.

Student Exploration: Hardy-Weinberg Equilibrium (ANSWER KEY)

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Student Exploration Calorimetry Lab Gizmo Answer Key

water. Calorimetry Lab Answers Correct Answer: A. Substance A A chemist mixes 500 g of lead at 500°C with 1,200 g of water at 20°C. She then mixes 500 g of copper at 500°C with 1,200 g of water at 20°C. The specific heat capacity of lead is 0.1276 J/g°C and the specific heat capacity of copper is 0.3845 J/g°C.

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Question: Student Exploration: Calorimetry Lab Vocabulary: Calorie, Calorimeter, Joule, Specific Heat Capacity Prior Knowledge Questions (Do These BEFORE Using The Gizmo.) 1. The Latin Word Calor Means "heat," And Meter Comes From The Greek Word Meaning To Measure." What Do You Think A Calorimeter Does? 2.

Student Exploration: Calorimetry Lab Vocabulary: C ...

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